

# Introduction to space-group symmetry



## 2022 Spring Festival Crystallographic School and Workshop on Crystal-field Applications

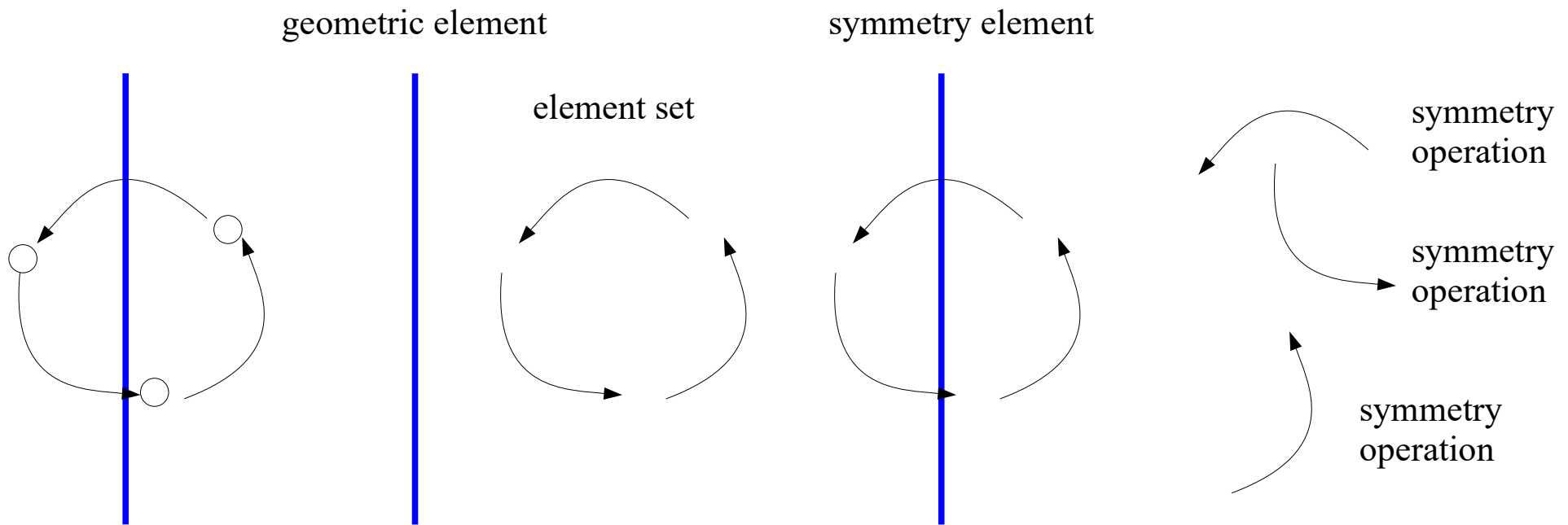
University of Science and Technology, Beijing, China,  
1-14 February 2022

Massimo Nespolo, Université de Lorraine, France  
[massimo.nespolo@univ-lorraine.fr](mailto:massimo.nespolo@univ-lorraine.fr)



# Elements and operations

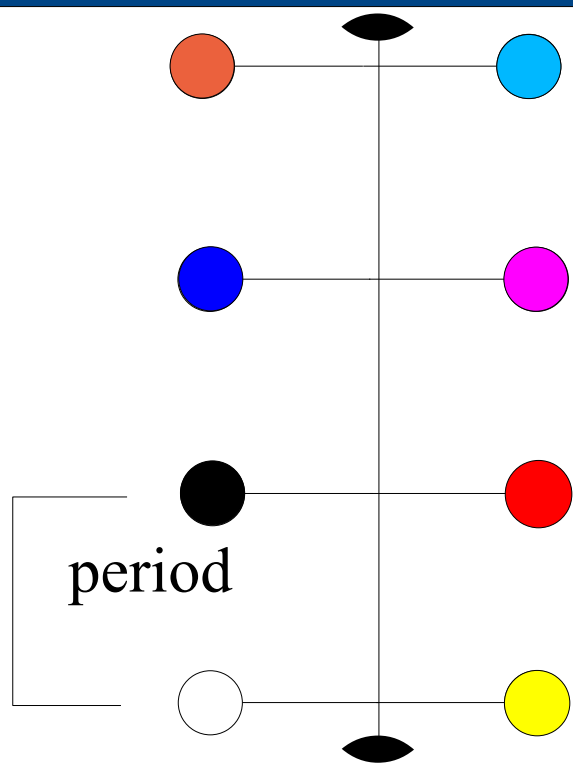
- **geometric element** : the point, line or plane left invariant by the symmetry operation.
- **symmetry element**: the geometric element defined above together with the set of operations (called **element set**) that leave it invariant.
- **symmetry operation** : an isometry that leaves invariant the object to which it is applied.



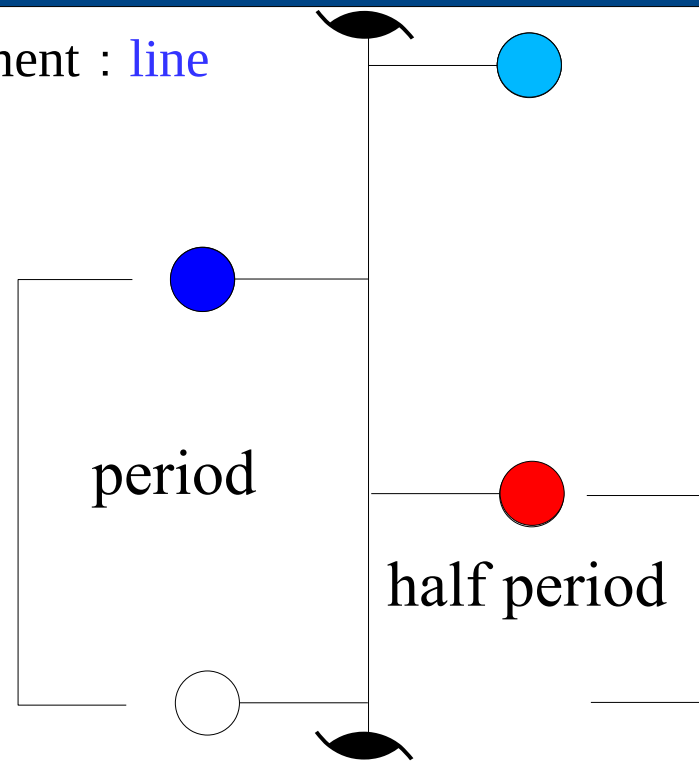
The operations that share a given geometric element differ by a lattice vector. The one characterized by the shortest vector is called **defining operation**.

# Screw axes $n_p$ (screw component: $p/n$ )

<http://dx.doi.org/10.1002/crat.201600129>



Geometric element : **line**



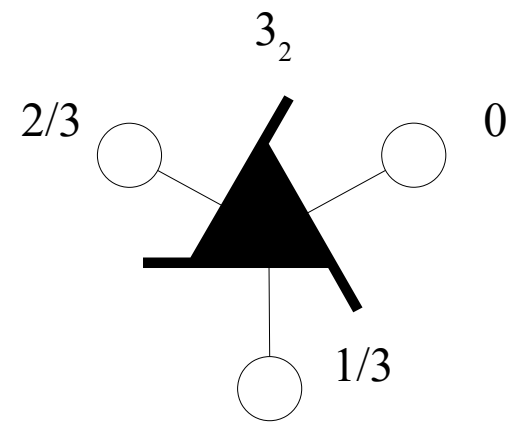
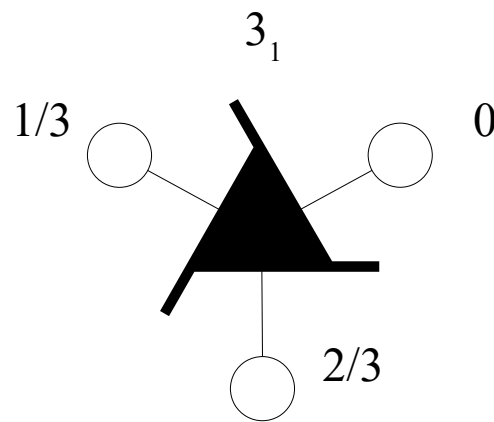
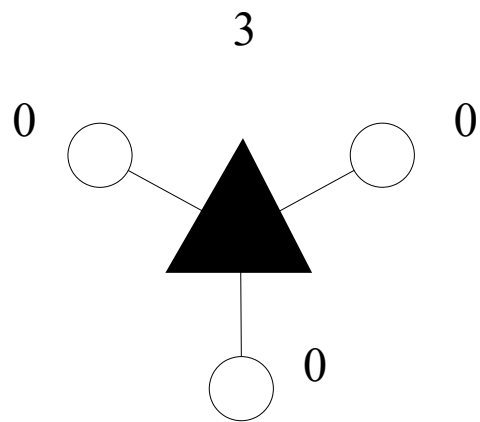
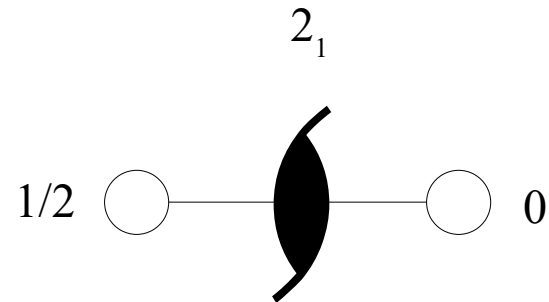
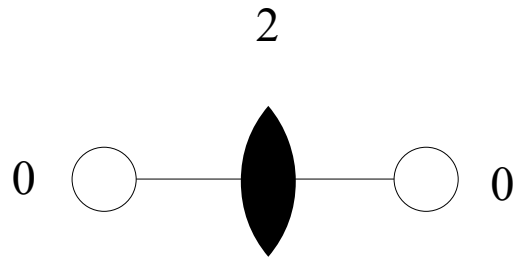
- $\rightarrow$   $2_0 = 2$
- $\rightarrow$   $t(001)$      $\rightarrow$   $2_2$
- $\rightarrow$   $t(002)$      $\rightarrow$   $2_4$
- $\rightarrow$   $t(003)$      $\rightarrow$   $2_6$

- $\rightarrow$   $2_1$
- $\rightarrow$   $t(001)$
- $\rightarrow$   $2_3$

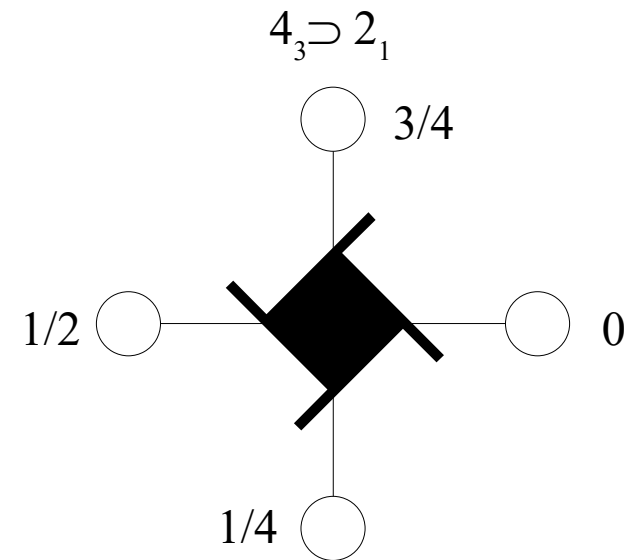
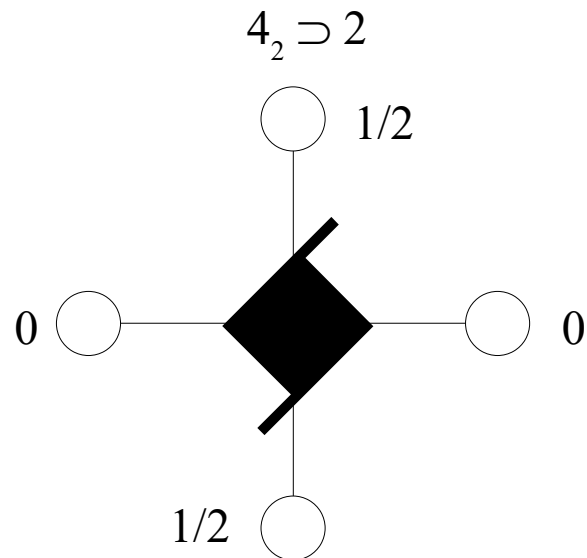
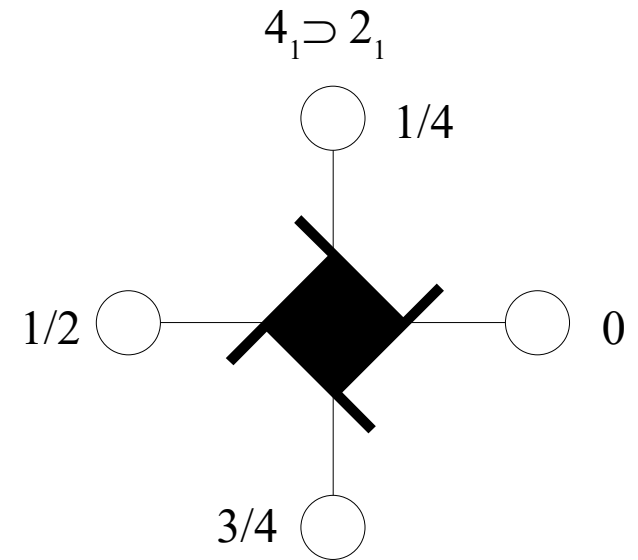
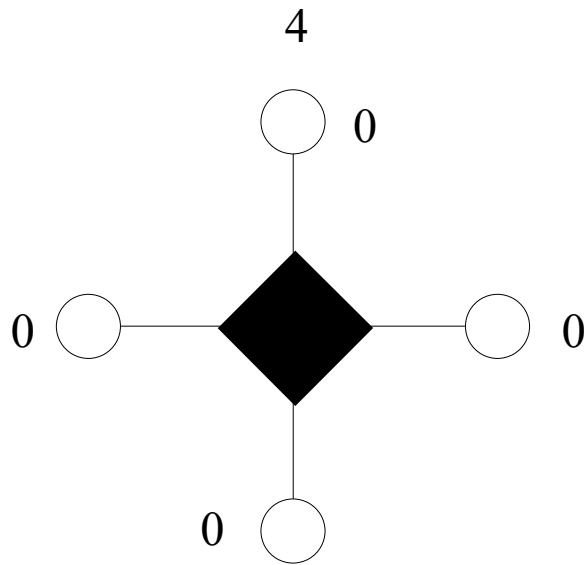
Defining operation : **twofold rotation**  
Symmetry element : **rotation axis**

Defining operation : **twofold screw rotation**  
Symmetry element : **screw axis**

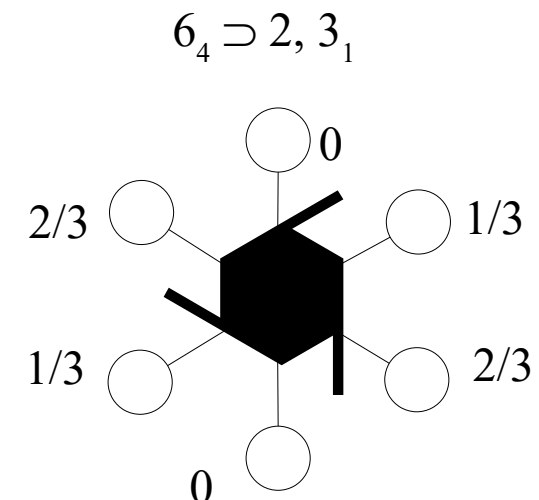
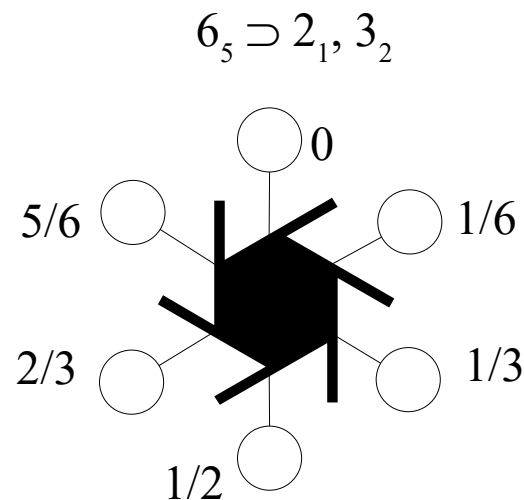
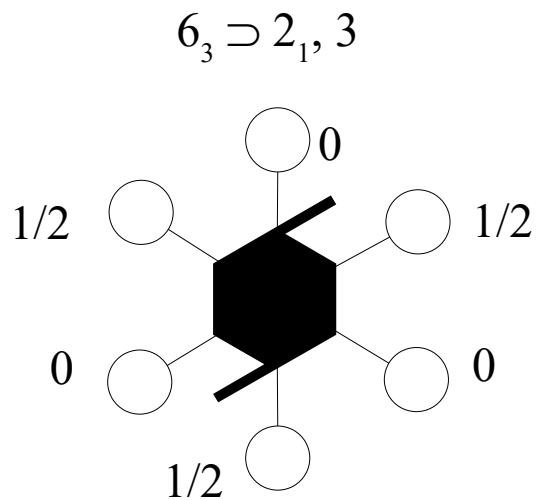
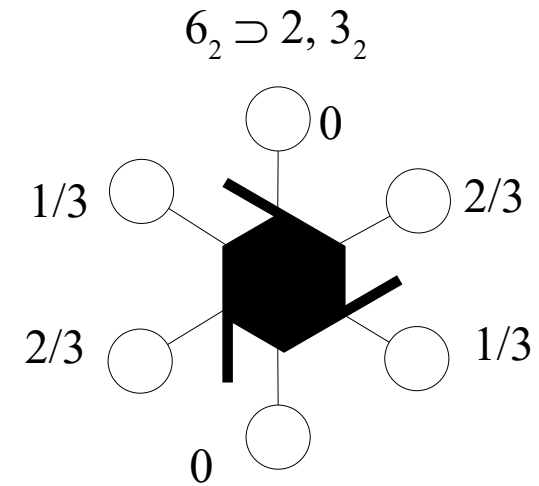
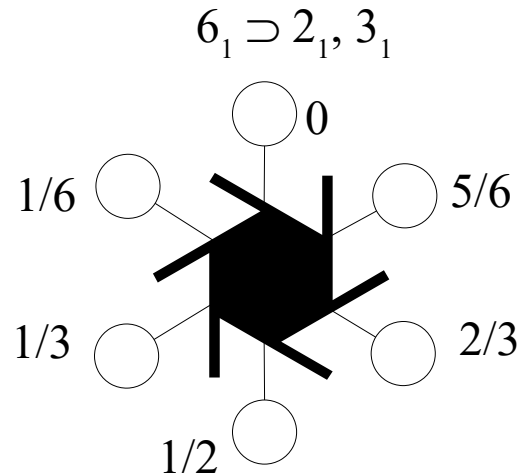
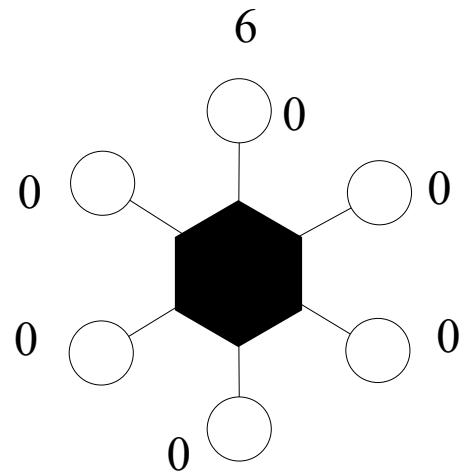
# Screw axes



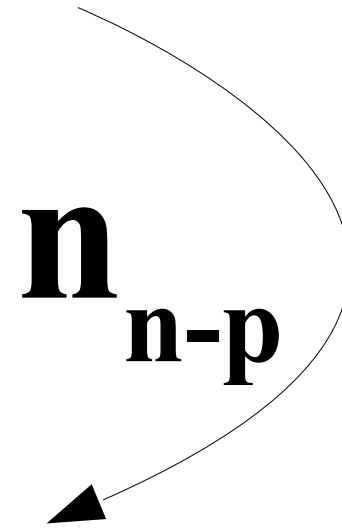
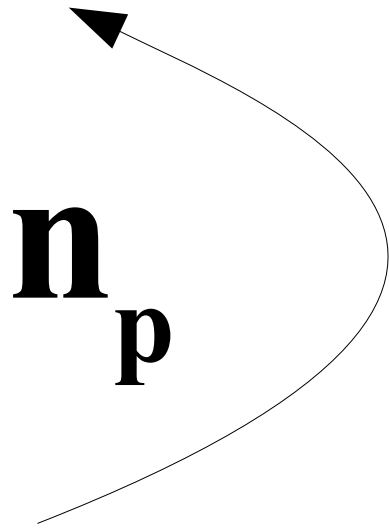
# Screw axes



# Screw axes



# Screw axes

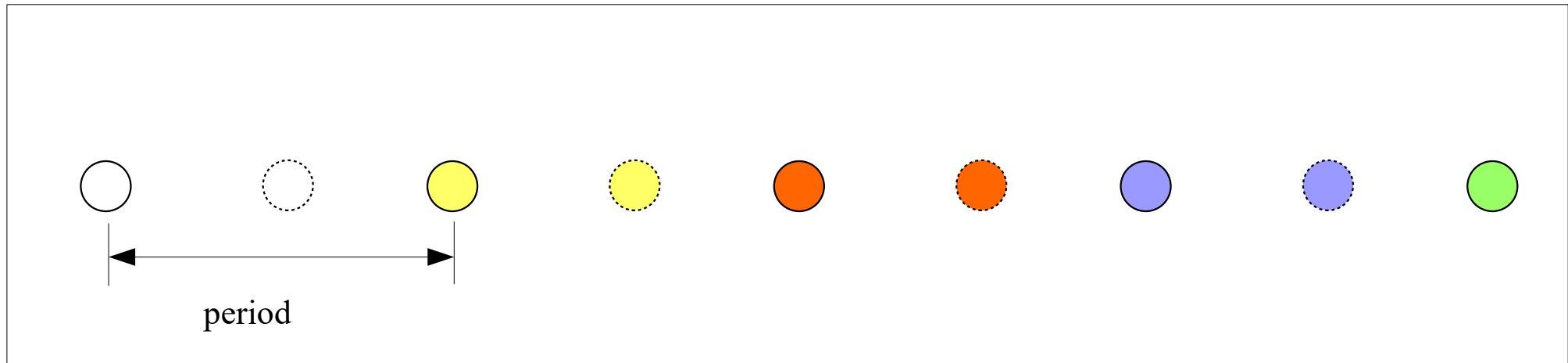


# Glide planes $g$

<http://dx.doi.org/10.1002/crat.201600129>

Geometric element : **plane**

↑ [001]



○ → ○  $g (\frac{1}{2}, \frac{1}{2}, 0) x, x, z$

○ → ●  $t(1, 1, 0)$

→ [110]

○ → ●  $g (\frac{3}{2}, \frac{3}{2}, 0) x, x, z$

○ → ●  $t(2, 2, 0)$

○ → ●  $g (\frac{5}{2}, \frac{5}{2}, 0) x, x, z$

○ → ●  $t(3, 3, 0)$

○ → ●  $g (\frac{7}{2}, \frac{7}{2}, 0) x, x, z$

○ → ●  $t(4, 4, 0)$

## Special cases

$g (0, 0, 0): m$

$g (0, \frac{1}{2}, 0): b$

$g$  (half-diagonal):  $n$

$g (\frac{1}{2}, 0, 0): a$

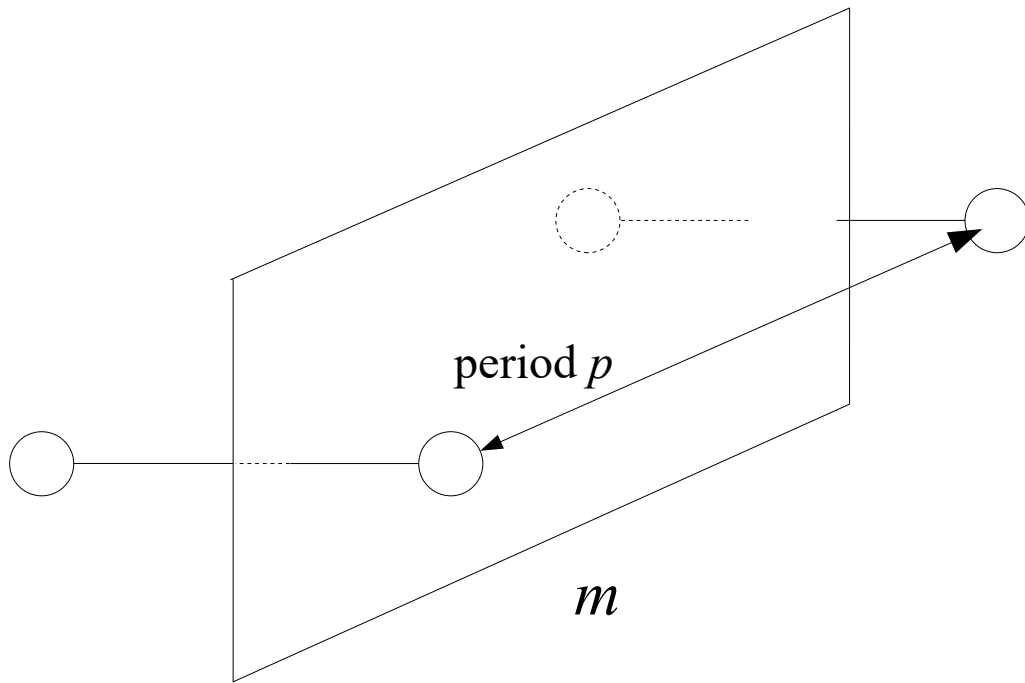
$g (0, 0, \frac{1}{2}): c$

$g$  (quarter-diagonal):  $d$

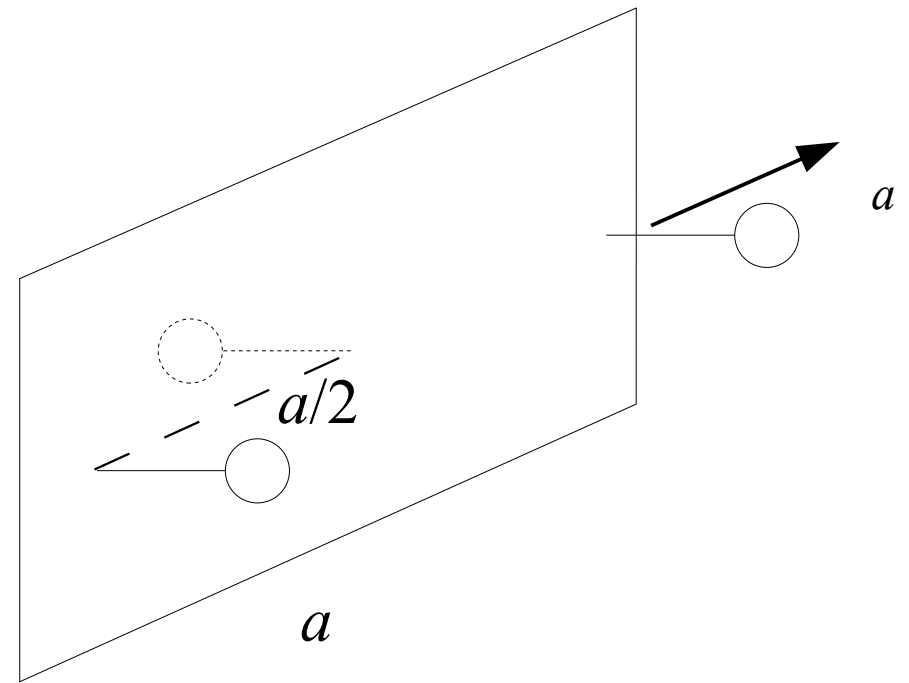


# Glide planes $g$

Geometric element: **plane**



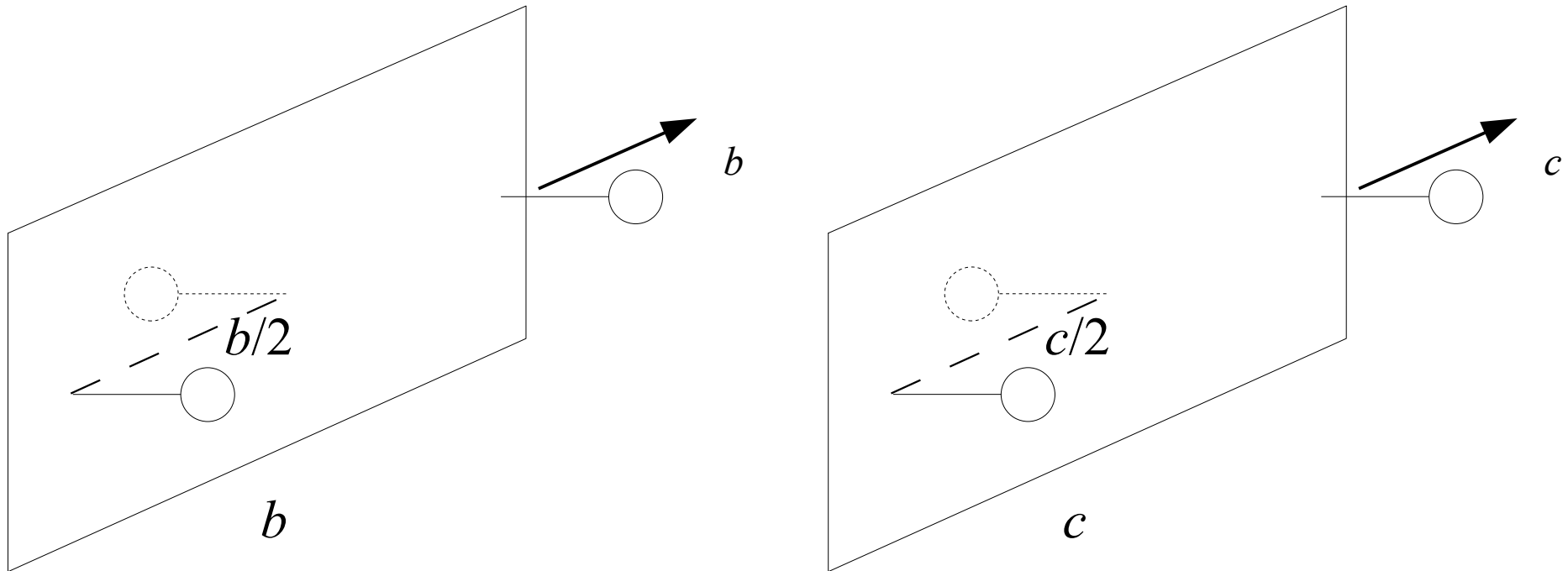
Defining operation: **reflection**  
Symmetry element: **mirror plane**



Defining operation: **glide reflection**  
Symmetry element: **glide plane**

# Glide planes $g$

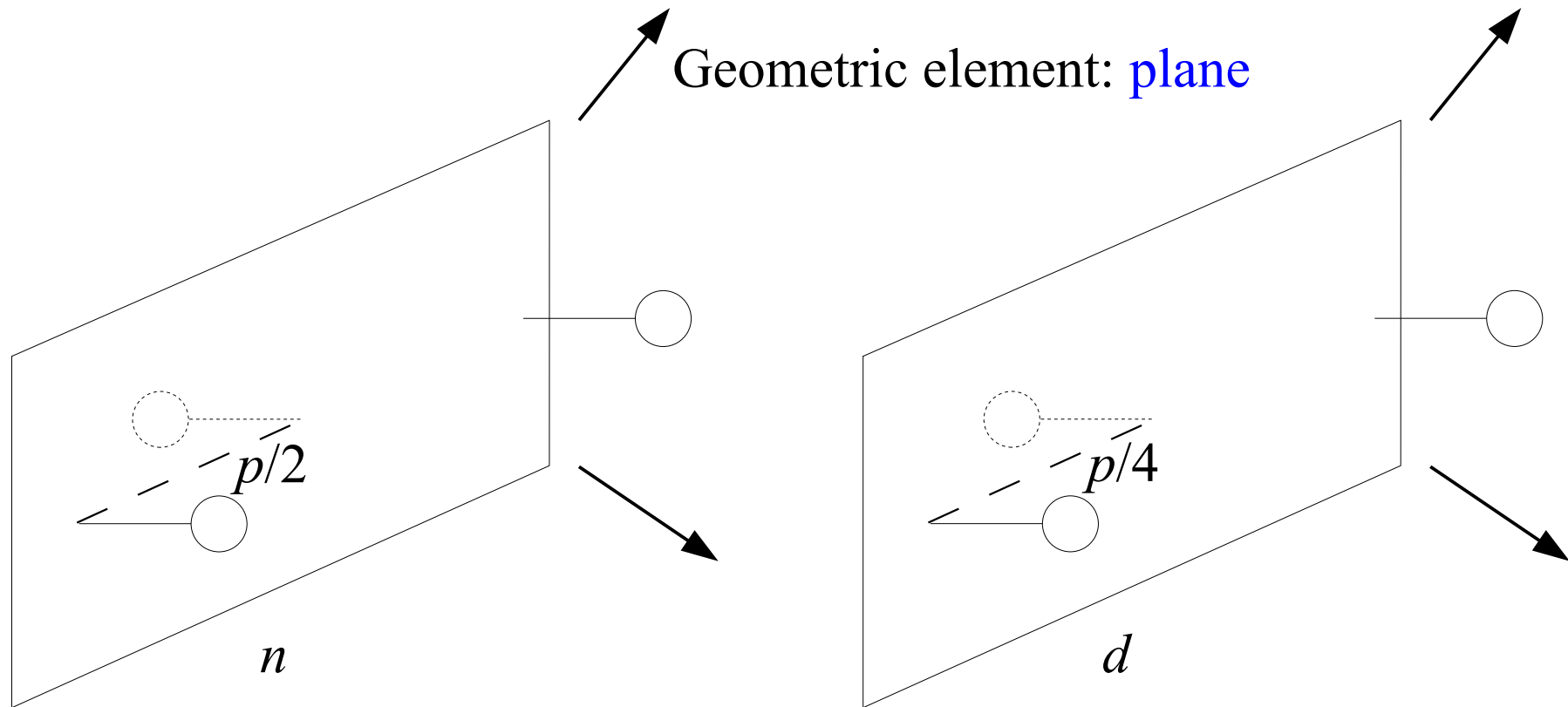
Geometric element: **plane**



Defining operation: **glide reflection**

Symmetry element: **glide plane**

# Glide planes $g$



Defining operation: glide reflection

Symmetry element: glide plane

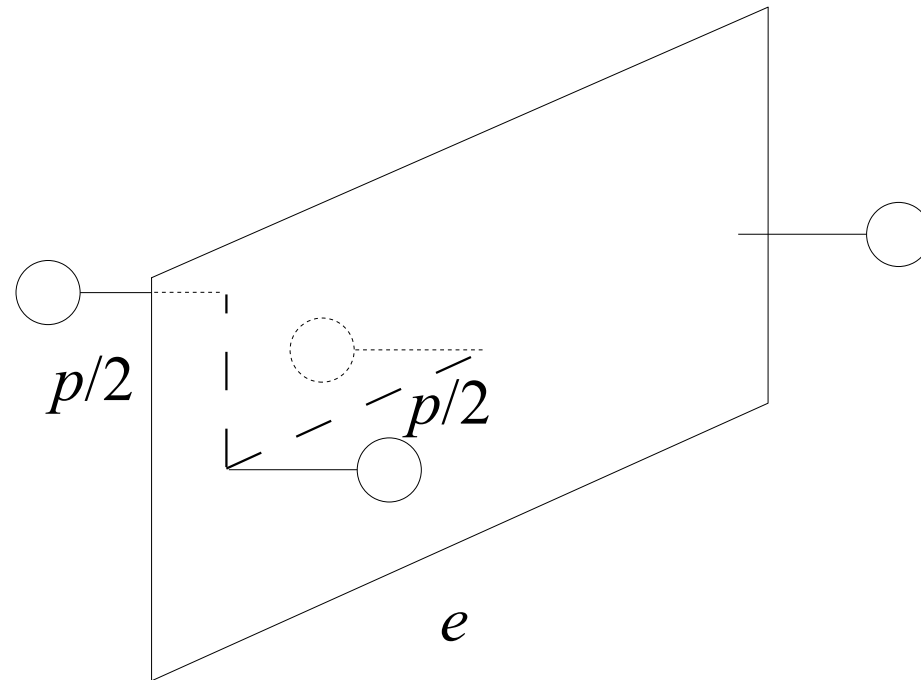
**diagonal**

Diamond structure glide plane

2D glide line; 3D glide plane with "unusual" glide

# Glide planes $g$

Geometric element: **plane**



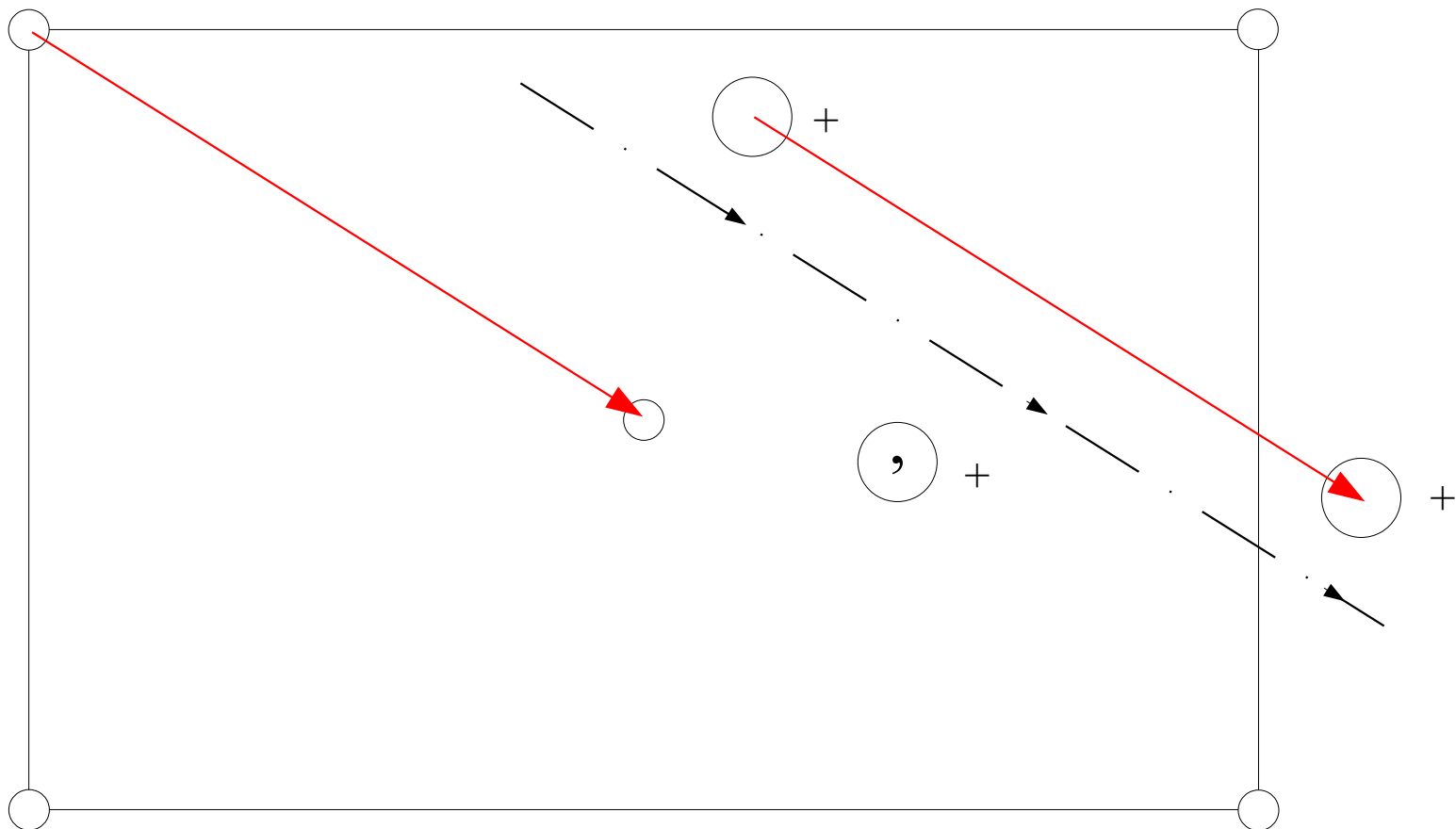
Defining operation: **glide reflection**

Symmetry element: **glide plane**

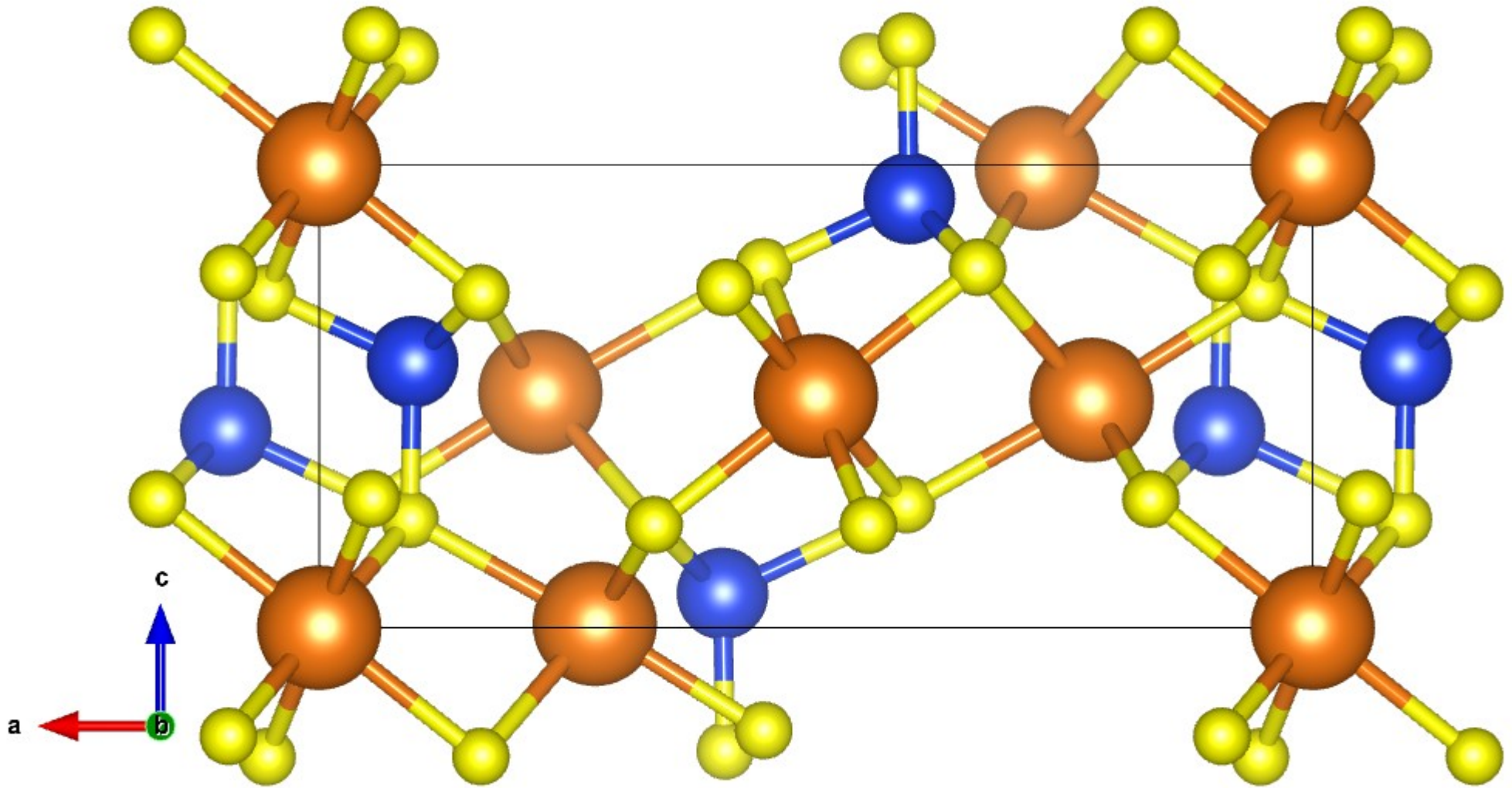
# How can we have a glide of $\frac{1}{4}$ if the reflection is an operation of order 2 ?

## If the unit cell is centred !

Vector centring  
the unit cell, with  
a norm  $p/2$

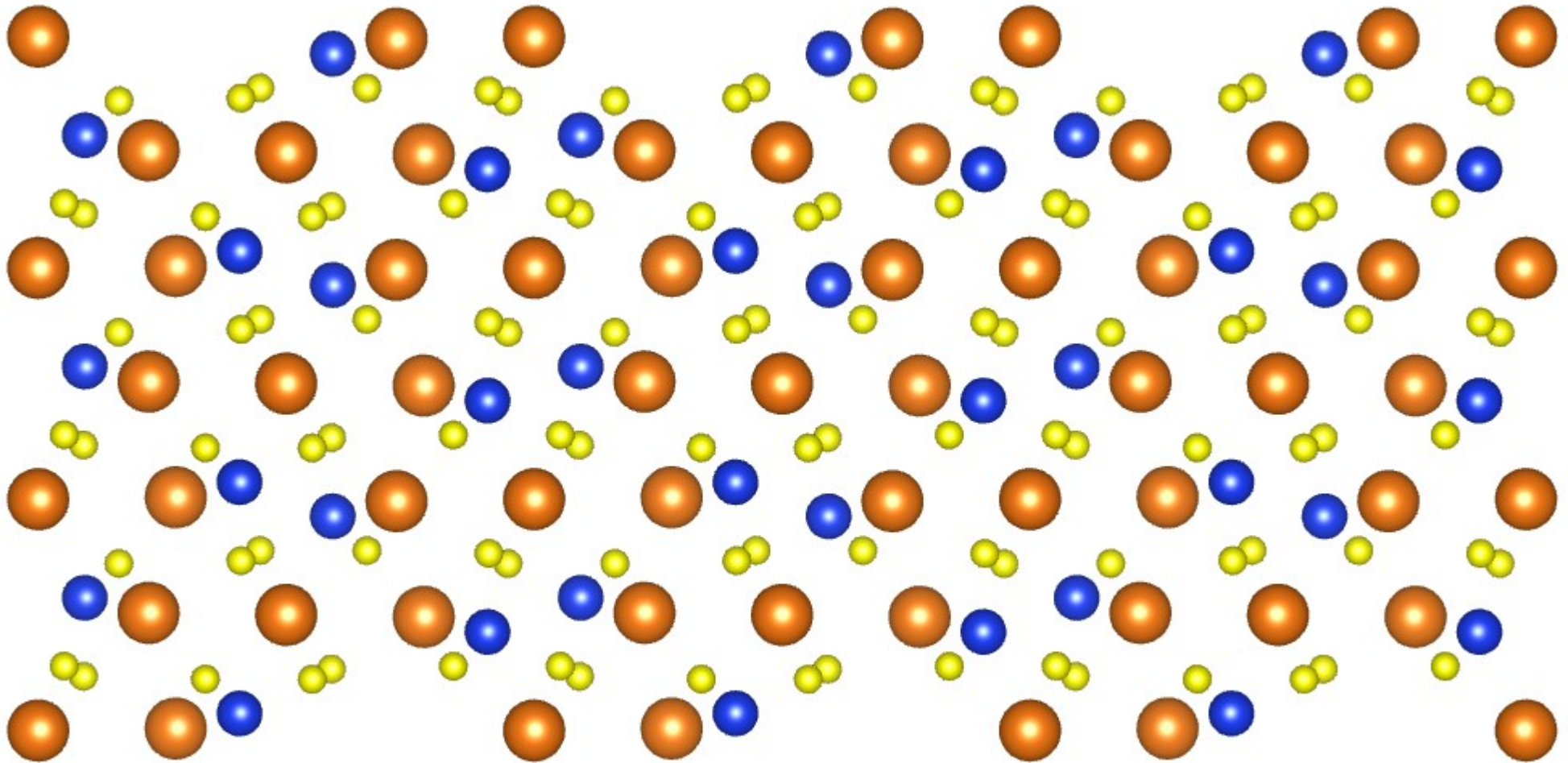


# Example of crystal structure



Same colour means same chemical species

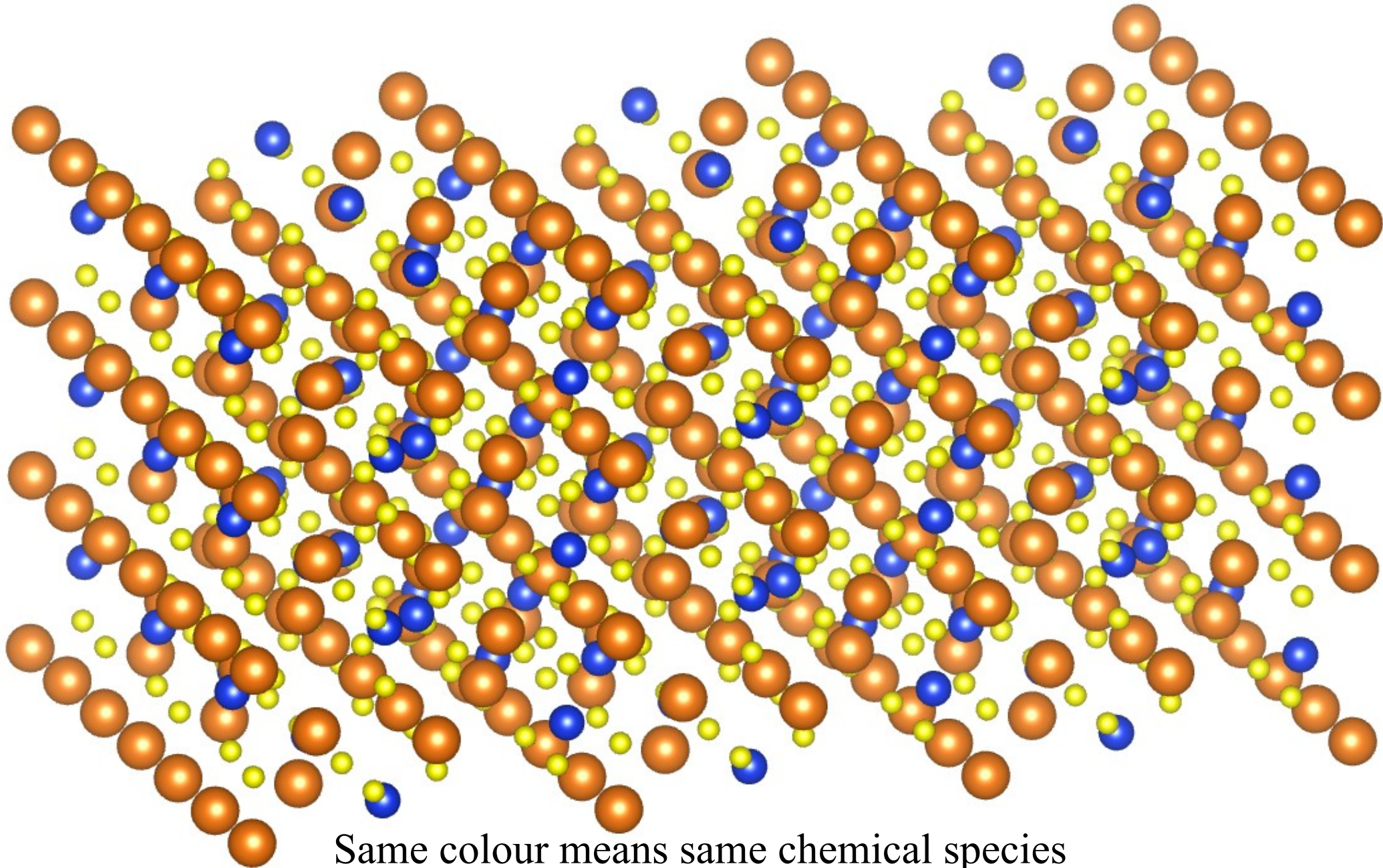
# Example of crystal structure



Same colour means same chemical species

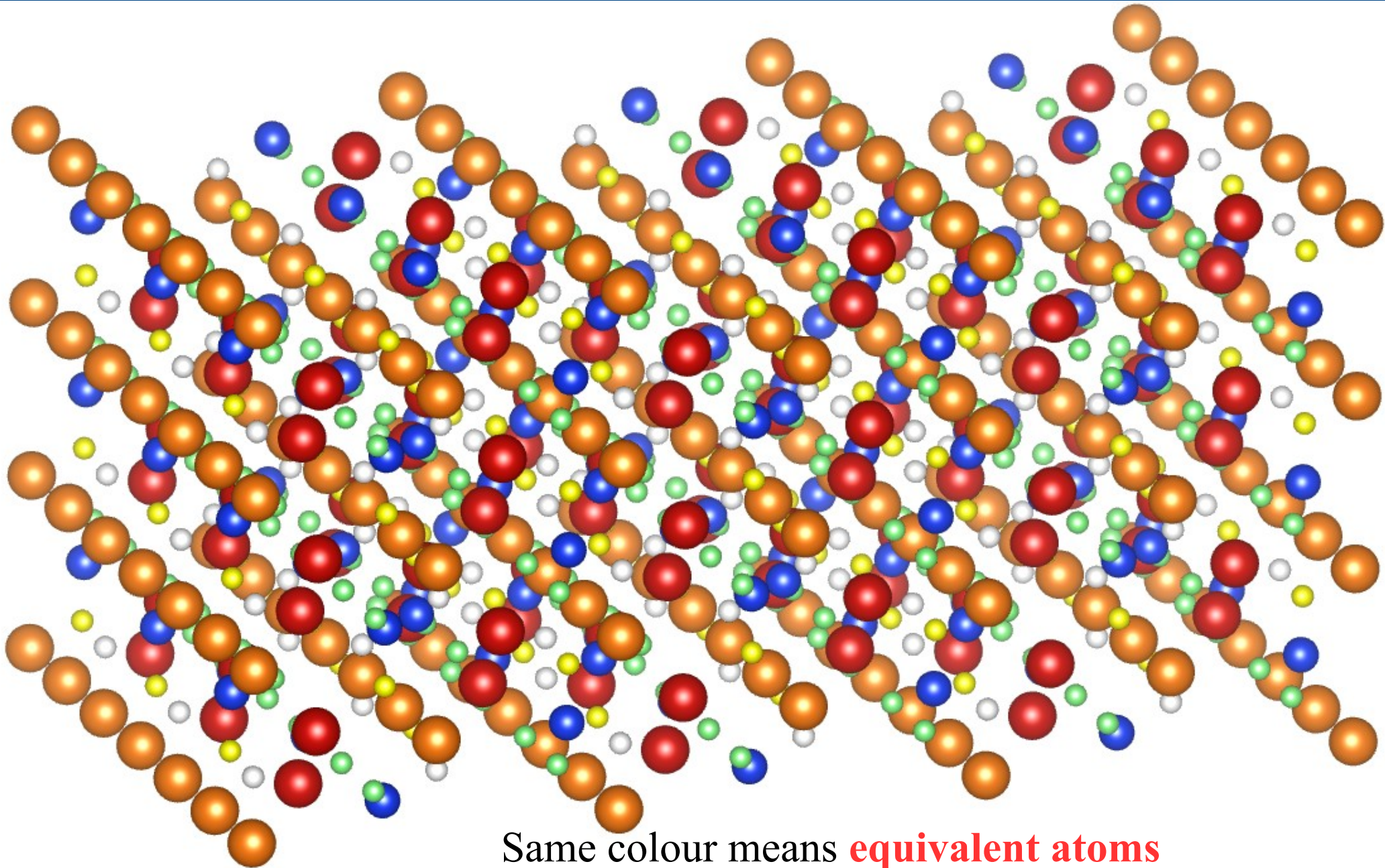


# Example of crystal structure





# Example of crystal structure



# Equivalent atoms

- The atoms of an infinite<sup>†</sup> set in a crystal structure are equivalent if they fulfil two conditions:
  - they all correspond to the same chemical species;
  - they can be related by global isometries.

Each set of atoms that fulfils the above conditions constitute **crystallographic orbit**.

The whole set of isometries that maps atoms belonging to the same crystallographic orbit constitutes the **eigensymmetry** of that orbit.

The whole set of atoms building up a solid<sup>‡</sup> can be divided into a finite number of crystallographic orbits. The solid is a crystal and the atoms define its crystal structure.

<sup>†</sup>The surface of a crystal is treated as a defect and ignored.

<sup>‡</sup>This definition ignores static and dynamic defects.

# Crystal structure, fractional atomic coordinates, crystallographic orbits, Wyckoff positions

**Crystal structure** atomic distribution in space that complies with the order and periodicity of the crystal

**Fractional atomic coordinates** : atomic coordinates  $x,y,z$  within a unit cell with respect to the basis vectors  $\mathbf{a},\mathbf{b},\mathbf{c}$ .

$\mathbf{a},\mathbf{b},\mathbf{c}$  (bold) : basis vectors

$a,b,c$  (italics) : reference axes and cell parameters

$x,y,z$  (italics) : fractional atomic coordinates

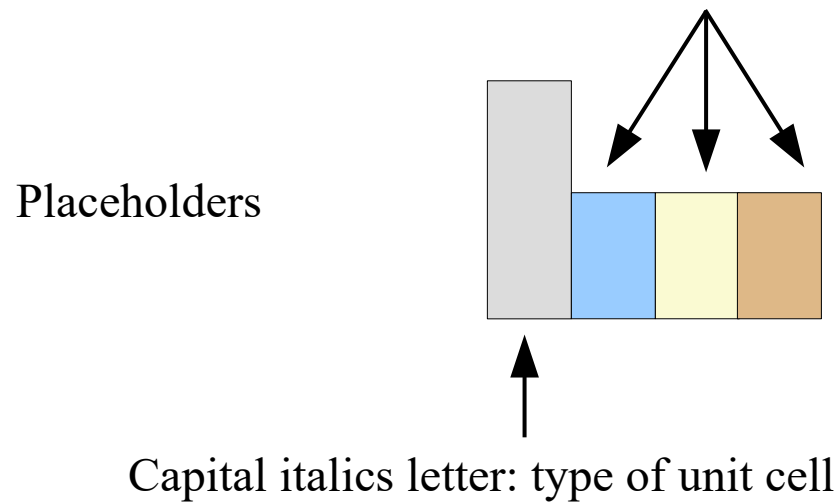
↓ Don't forget the translations!

**crystallographic orbit** : the infinite set of atoms obtained by applying all the symmetry operations of the space group to a given atom in the unit cell.

**Wyckoff positions** : classification of the crystallographic orbits on the basis of the symmetry of the atomic positions (site-symmetry group) ( N to 1 mapping )

# Hermann-Mauguin symbols for space groups

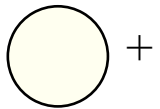
**Up to** three numbers or low-case letters (*italics*): symmetry **elements** along the symmetry directions of the lattice



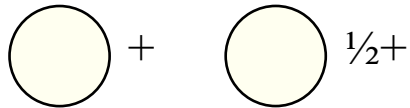
Example:  $C2/c$ ,  $Pbam$ ,  $I4_1/acd$ ,  $Fd\bar{3}c$  etc.

# Graphical symbols for atoms projected in the unit cell

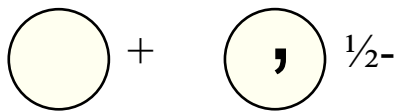
## Direction of projection $c \rightarrow$ vertical coordinate $z$ .



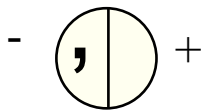
An atom with coordinate  $z > 0$  (“+”).



Two atoms mapped by an operation of the first kind (handedness preserving operation) with coordinates  $z (> 0)$  and  $\frac{1}{2}+z$  respectively.

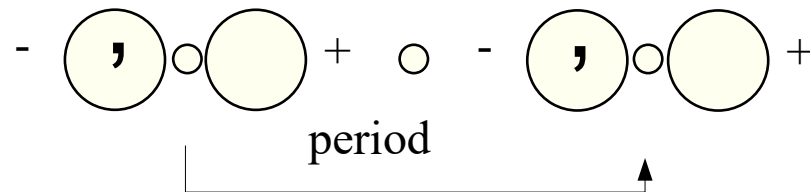
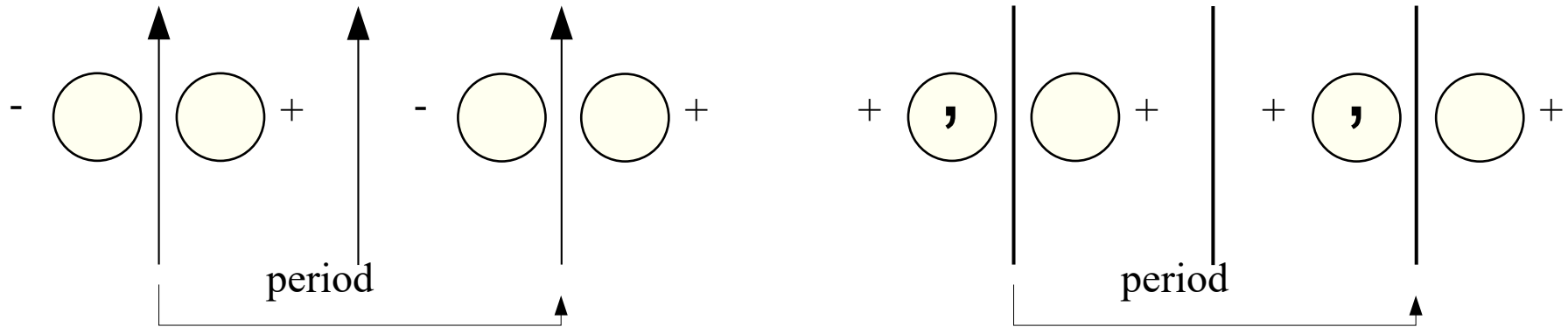


Two atoms mapped by an operation of the second kind (handedness reversing operation: note the “comma”) with coordinates  $z (> 0)$  and  $\frac{1}{2}-z$  respectively.



Two atoms mapped by an operation of the second kind (handedness reversing operation: note the “comma”) with coordinates  $x,y,z$  and  $x,y,\bar{z}$  respectively, overlapped in projection. The vertical segment represents a “cut” of the atom above (left) which allows to see half of the atom below (right).

# Symmetry elements corresponding to involutions occur every half period





# Reading information off the Hermann-Mauguin symbols for space groups

Procedure to read the symbol:

remove the capital letter  
replace screw axes  $n_p$  (if any) with rotation axes  $n$   
replace glide planes (if any) with mirror planes  $m$

Obtain the point group of the space group

Obtain the crystal system and the conventional unit cell

Interpret the symbol of the space group

Example:  $C2/c$ ,  $Pbam$ ,  $I4_1/acd$ ,  $Fd\bar{3}c$  etc.

# Site-symmetry groups (stabilizers) and Wyckoff positions of space groups

Let  $G$  be a space group (thus, **infinite**) and  $X$  a point in space.

The **infinite** set of points  $\{GX\} = \{X, X', X'' \dots\}$  is the **crystallographic orbit** of  $X$  under the action of  $G$ .

A finite subgroup  $S$  of  $G$  ( $S \subset G$ ) leaves  $X$  invariant, *i.e.* it “stabilizes”  $X$ :  $SX = X$

$S$  is called the **site-symmetry group** (or **stabilizer**) of  $X$ .

Crystallographic orbits for which  $S = \{1\}$  are called **general orbits**.

Crystallographic orbits for which  $S \supset \{1\}$  are called **special orbits**.

Two points  $X$  and  $X'$  belonging to the same crystallographic orbit, whose site-symmetry groups  $S$  and  $S'$  are conjugate under  $G$ , belong the same class, which is known as **Wyckoff position**.

$$s \in S, sX = X \quad s' \in S', s'X' = X' \quad g \in G, gX = X' \quad s' = gsg^{-1}$$

$X, X'$  belong to the same Wyckoff position.



# Site-symmetry groups (stabilizers) and Wyckoff positions of space groups

The number of points belonging to a crystallographic orbit is infinite.

The number of points of a given crystallographic orbit within a single unit cell is known as the **multiplicity** of the Wyckoff position.

The multiplicity  $M$  of the **general** Wyckoff position is obtained as:

$$M(\text{general}) = O(P) \times M(U)$$

Order of the point group  $P$  of  $G$

Multiplicity of the unit cell  $U$  of  $G$

The multiplicity  $M$  of a **special** Wyckoff position is obtained as:

$$M(\text{special}) = M(\text{general}) / O(S)$$

Order of the site-symmetry group of the special position